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Impact of aluminum concentration on the synthesis of highly pure Cr₂AlC MAX phase

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In recent years, MAX phases achieved significant attention from scientists for their promising practical applications and theoretical values. This fascinating class of materials are layered ternary carbides and nitrides having hexagonal crystal structure. MAX phases exhibit a unique combination of properties of both metals and ceramics such as high thermal and electrical conductivity, excellent machinability, good damage tolerability and superb thermal shock resistance, as compared to binary carbides and nitrides. However, the wide applicability of MAX phases restricts due to cost effective synthesis and their phase stability. In this study, we have successfully synthesized Cr₂AlC MAX phase through pressure less sintering route. The impact of aluminum content (10 – 50 mol%) on the phase formation and structural properties of Cr₂AlC has been investigated. It is observed that the purity of Cr₂AlC MAX phase can be enhanced by varying the aluminum content, even at lower temperatures (1200–1300°C), as compared to previously reported Cr₂AlC MAX phase (> 1300°C). The highly pure Cr₂AlC is obtained at 1300°C, when the aluminum content in 40 mol%. In addition, the crystallite size is found maximum (28.14 nm) and lattice strain (4.03×10^{-3}) is minimum for highly pure Cr₂AlC. Also, oxidation kinetic analysis is performed to estimate activation energy (E_g) by following Kissinger-Akahira-Sunose (KAS) method. The MAX phase demonstrated high oxidation resistance and no mass gain is observed in the TGA curve below 700°C. In addition, two exothermic peaks are observed at ~ 700°C and ~ 1050°C. The former peak is associated the oxidation of excess aluminum present on the surface of MAX phase. While later peak represents the formation of Al₂O₃ layer over the surface MAX phase.

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