

## Low-temperature Heat Treatment (80°C) Effect on the Electrochemically Synthesized $\text{CuInTe}_2$ Thin Films for Energy Harvesting Applications

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### Abstract

We have investigated the effect of low-temperature (80°C) post-deposition heat treatment onto  $\text{CuInTe}_2$  (CIT) thin films prepared by one-step electrochemical technique. Aqueous precursor solution consisting ionic species of Cu, In and Te with pH 4.0 was used for the growth of CIT layers. The deposition potential was optimized using cyclic voltammetry. Conventional three-electrode geometry was used to electrodeposit CIT thin films onto CdS substrates. The structural, optical, morphological, compositional and transport properties were studied with the aid of XRD, Raman, HRTEM, UV-Visible, FESEM, EDAX, I-V and C-V measurements. As-deposited samples were amorphous in nature, however upon heat treatment highly crystalline CIT thin films with tetragonal crystal structure were exhibited. The values of energy band gap for the films deposited at -0.7 V and -0.8 V versus Ag/AgCl were estimated to be in the range 1.02 eV to 1.1 eV. Compact, uniform, void free and well adherent films were deposited at -0.7 V and -0.8 V. The samples were heat treated at 80°C for 60 minutes therefore, visible changes in the surface morphology were not observed by SEM. Indium rich films were electrodeposited for -0.7 V and -0.8 V, however upon heat treatment stoichiometric layers were obtained. Schottky diodes are formed with Au metal contact in all cases. The solar cell investigated under illumination at 1.5 AM exhibits the short circuit current density ( $J_{sc}$ ), 40.75 mA/cm<sup>2</sup>; open circuit voltage ( $V_{oc}$ ), 255 mV; fill factor (FF), 41% and power conversion efficiency ( $\eta$ ), 4.01%. This low-temperature heat treatment procedure could be advantageous for the fabrication of CIT solar cells onto flexible substrates.

**Keywords:**  $\text{CuInTe}_2$ ; Electrodeposition; Energy harvesting; Thin films; Solar cells

### Introduction

The ternary chalcopyrite semiconductors are technologically important group of semiconductors because of their photo-functionality. The broad range of optical band gap offered by these materials has led their emergence as technologically significant device materials in photovoltaic industry. Recently, researchers at Stuttgart's Centre for Solar Energy and Hydrogen Research Laboratory have reported highest efficiency 21.7% on CIGS material [1]. The material's growth and processing on different substrates showed major changes in the material properties because of the change in geometry/orientation of the substrate [2]. Therefore, it is important to investigate the material properties on different interfaces which lead the researchers to garner a fundamental understanding of materials as well as device perspective. The influence of growth parameters and subsequent post-deposition heat treatment are equally important to get the preferred material with device properties. Duchatelet [3] have reported 12.4% efficiency from electrodeposited CIGS with annealing at 500 – 550°C under the flow of pure hydrogen and subsequently sulfurization at 550 – 600°C. Kobayashi [4] have reported the growth of CIGS by molecular beam epitaxy technique with three stage procedure with substrate 540°C. Mise and Nakada [5] have thoroughly studied the effect of substrate temperature and film thickness on properties of  $\text{CuIn}_3\text{Te}_5$  thin films prepared by single step co-evaporation. The films deposited at 250°C temperature measured cell efficiency 6.28%. Subsequently the same group reported 6.92% efficiency [6] in presence of  $\text{MgF}_2$  antireflection coating ( $\text{MgF}_2$ ).

$\text{CuInTe}_2$  thin films can be synthesized by vacuum as well as non-vacuum techniques. Numerous preparation techniques have been reported in the literature, viz, co-evaporation of Cu, In and Te by molecular beam epitaxy [5,6], Bridgeman method [7], thermal evaporation [8], flash evaporation [9], chemical vapor transport [10] and electrodeposition [11]. The synthesis of CIT nanoparticles [12] and quantum dots [13] are also reported. The highest efficiency 6.92% has

been reported by molecular beam epitaxy [6] technique with  $\text{CuIn}_3\text{Te}_5$  as absorber layer. The substrate temperature 250°C was maintained during deposition for base pressure  $4 \times 10^{-6}$  Pascal. The reported efficiency is still low could be due to recombination via trap state in bulk  $\text{CuIn}_3\text{Te}_5$ , poor quality of the film near the substrate and band gap fluctuations because of the disorders. CIT films have also been prepared by Bridgeman methods at high temperature  $\sim 1100^\circ\text{C}$  [7]. The films deposited by thermal evaporation and graphite box annealing contain number of secondary phases of  $\text{In}_x\text{Te}_y$  and  $\text{Cu}_2\text{Te}$  [8]. The single phase polycrystalline and single crystalline films have been obtained by flash evaporation and chemical vapor transport, respectively. The band gap for single crystalline and polycrystalline CIT films can vary from  $\sim 0.91$  eV to 1.00 eV [14,15]. Electrodeposition is one of the low-cost and very successful techniques used for the deposition of chalcopyrite and kesterite thin films for photovoltaic applications [14-22], however, limited literature is available on the synthesis of CIT by electrodeposition technique [11,20,23,24]. Furthermore, none of the group has developed the solar cell devices with the electrodeposited CIT. Some limitations of electrodeposition technique are still open to resolve, such as the requirement of conducting substrate and the growth of controlled stoichiometric thin film over large area. The difficulty in growth of stoichiometric film is with electrodeposition are still open for thinner layer probably due to the uneven distribution of current.

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This issue can be resolved with some extent for thicker film of thickness 2 to 3 μm. Here, we report the electrodeposition conditions for CIT and the effect of low-temperature heat treatment was thoroughly investigated by various characterization techniques. Low processing temperature of solar cell demonstrates good device properties, which could be effectively used for the preparation of solar cells on flexible substrates [25].

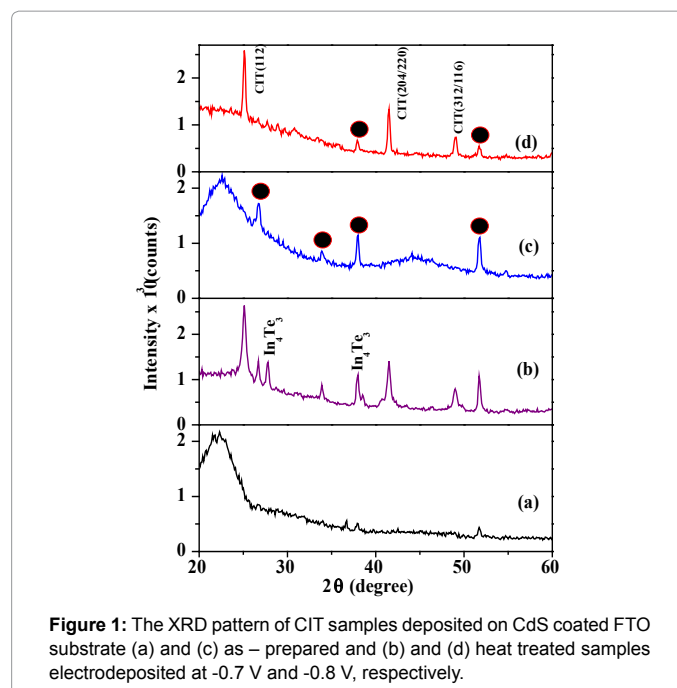
## Experimental Details

CuInTe<sub>2</sub> thin films were potentiostatically electrodeposited on CdS coated conducting glass substrates at 75°C from aqueous bath for pH ~ 4. CdS films were deposited onto fluorine doped tin oxide (FTO) coated glass substrates by chemical bath deposition. The preparation and characterization of the thin films is reported elsewhere [26]. Potentiostat/galvanostat Model, Biologic SP 300 was employed to study the cyclic voltammetry (CV) experiment and the electrodeposition of CIT films. The detailed growth mechanism along with half-electrode reaction is reported in our previous work [25]. CIT films were electrodeposited for -0.7 V and -0.8 V versus Ag/AgCl reference electrode. Prior to heat treatment the samples were stored in laboratory atmosphere for overnight to remove the water contents. Subsequently, the samples were heat treated at 80°C for an hour with heating rate 1°C per minute and cooled down to room temperature naturally. The electrodeposition of the sample was performed several times (10 to 20 times) in the freshly prepared electrolyte in order to study the adherence of layer to the substrate and compactness. The heat treatment experiment at 80°C, was performed onto CIT layers as well as CdS/CIT hetero-structure sole cell device at least 10 different sample to study the repeatability and reproducibility of the results. Typical results obtained on CIT and CdS/CIT solar cells are reported. Structural, optical, morphological, compositional, electrical and optoelectronic properties were studied thoroughly. The heat treatment results were repeated more than 4/5 times and results were repeatable. X-ray diffractometer (Model Bruker D8 Advance, Germany) of Cu Kα radiation, with λ = 0.154 nm was used to study the structural properties. Invia Renishaw Raman Microscope coupled with Philips CCD camera was employed to study the Raman spectroscopy. The continuous wave laser of wavelength 785 nm was used as an excitation source. The HRTEM and diffraction pattern was examined using TEM, model TECHNAI G<sup>2</sup>. Optical measurements were performed using JASCO, UV-VIS-NIR Spectrophotometer model V-670. Elemental composition was obtained by TEM instrument. High magnified topographical images were obtained using FESEM, Model HITACHI S4800. Thermal evaporation system was employed to make the metal contacts on CIT samples for electrical and optoelectronic measurements. Frequency response analyzer (FRA) facility available in above mentioned potentiostat was used to study the C-V measurements at frequency 100 KHz. The transport properties were studied by ECOPIA HMS-3000 model at constant magnetic field 0.54 T and probe current 10 mA. The surface of CIT layer was etched with NaCN solution (0.2%) for 90 seconds prior to metal contact. The dark and illuminated J-V measurements were performed using above potentiostat equipped with necessary software and probe system at light intensity 100 mW cm<sup>-2</sup> (AM 1.5).

## Results and Discussion

### X-Ray Diffraction

XRD spectra recorded for the as-deposited and heat treated samples electrodeposited for potentials -0.7 V and -0.8 V is depicted in Figure 1. Upon heat treatment at 80°C for an hour, the sample exhibited polycrystalline nature, whereas untreated samples were



**Figure 1:** The XRD pattern of CIT samples deposited on CdS coated FTO substrate (a) and (c) as – prepared and (b) and (d) heat treated samples electrodeposited at -0.7 V and -0.8 V, respectively.

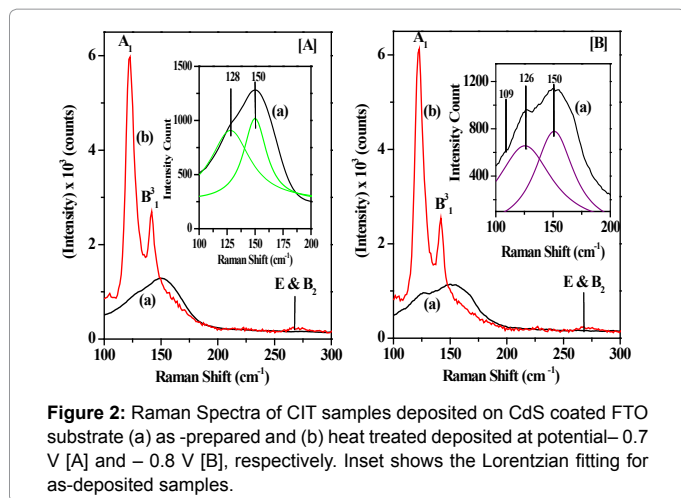
amorphous. The sharp XRD peaks attributed from heat treated samples at  $2\theta = 24.85^\circ$ ,  $41.09^\circ$  and  $48.54^\circ$  are corresponds to reflections (1 1 2), (2 0 4) / (2 2 0) and (3 1 2) / (1 1 6) respectively. CIT sample deposited at -0.8 V upon subsequent heat treatment exhibited the peak related to tetragonal structure of CIT along with the FTO (substrate) peaks, which are marked as solid circle (•). Additionally, secondary phases of In<sub>4</sub>Te<sub>3</sub> along with CIT reflections were observed in the sample deposited at -0.7 V upon thermal heat treatment. The crystallinity for the CIT sample deposited at -0.8 V after heat treatment was found to be higher as compare to the sample deposited at -0.7 V after heat treatment. The average crystallite size is calculated by the Debye Scherrer equation [25]

$$D = \frac{0.9\lambda}{\Delta(2\theta) \cos\theta} \quad (1)$$

where D is the average crystallite size, λ is the X-ray wavelength (1.54056 Å), θ and Δ(2θ) are the Bragg diffraction angle of the XRD peak in degree and the full width at half maximum in radian, respectively. The average crystallite size 20 nm and 28 nm were determined for the heat treated samples deposited at -0.7 V and -0.8 V, respectively. The low-temperature heat treatment phenomenon related to enhancement in the crystallinity could be explained by lithification, diagenesis/catagenesis and metamorphism [27,28].

### Raman Spectroscopy

Figures 2a and 2b depicts the Raman spectra of pristine and heat treated CIT samples electrodeposited for potentials -0.7 V and -0.8 V, respectively. The peaks observed in Raman spectra are in good agreement with the reported literature values [29]. In untreated samples, the broad hump from 100 cm<sup>-1</sup> to 175 cm<sup>-1</sup> is due to the mixing of 109 (B<sub>2</sub>), 122 (A<sub>1</sub>), 127(E), 141 (B<sub>1</sub>), 159 (E) and 170 cm<sup>-1</sup> (E and B<sub>2</sub>) modes which is in good agreement with IR data [29]. The B<sub>2</sub> (109) mode is silent in the sample deposited at -0.7 V. Sharp peaks are observed upon annealing corresponds to A<sub>1</sub>, B<sub>1</sub><sup>3</sup> and combination of E and B<sub>2</sub> modes. The related details are discussed elsewhere [25,29]. The spectra recorded for heat treated samples deposited at -0.7 V and -0.8



**Figure 2:** Raman Spectra of CIT samples deposited on CdS coated FTO substrate (a) as -prepared and (b) heat treated deposited at potential - 0.7 V [A] and - 0.8 V [B], respectively. Inset shows the Lorentzian fitting for as-deposited samples.

Mode	Standard Mode frequencies (cm <sup>-1</sup> )	Observed mode frequencies (cm <sup>-1</sup> )			
		-0.7 V		-0.8 V	
		Untreated	Heat treated	Untreated	Heat treated
E	109	silent	silent	109	silent
A <sub>1</sub>	123	silent	123	silent	123
E	128	128	silent	126	silent
B <sub>3</sub> <sup>1</sup>	143	silent	142	silent	142
E	159	155	silent	155	silent
E <sup>5</sup> and/or B <sub>2</sub> <sup>3</sup>	171	180	silent	silent	silent
E and B <sub>2</sub>	267	261	267	silent	266

**Table 1:** Phonon mode frequencies of CIT samples on CdS coated FTO substrate compared with standard literature data.

V were very similar. The phonon modes attributed in the CIT samples and the literature values are summarized in Table 1.

### Transmission Electron microscopy (TEM)

The high resolution transmission electron microscopy (HRTEM) and selected area diffraction pattern of the untreated and heat treated CIT films are shown in Figures 3a and 3b. The HRTEM images exhibits clear lattice fringes in heat treated samples (b) and (d) deposited at -0.7 V and -0.8 V, respectively. However the lattice fringes were not observed in the untreated samples. The interplaner distance ‘d’ measured for the samples deposited at -0.7 V and -0.8 V was found to be 0.19 nm and 0.36 nm which corresponds to the interplaner spacing of plane (316) and (112) of chalcopyrite CIT. The sharp continuous rings indicate the polycrystalline nature. These results agree well with structural results obtained by X-ray diffraction.

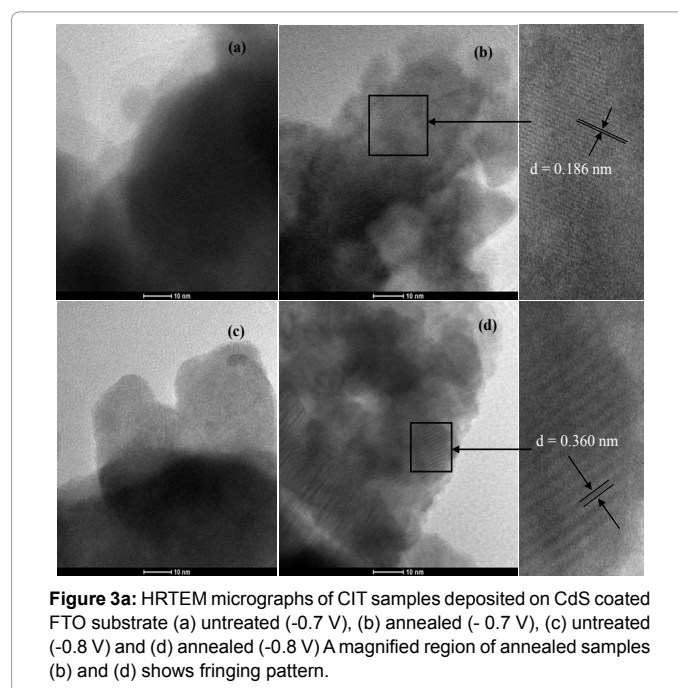
### Optical properties

Figures 4a and 4b shows the plots of (αhν)<sup>2</sup> verses energy (hν) of CIT thin films deposited at -0.7 and -0.8 V respectively. The data measured for untreated and heat treated sample is marked as (a) and (b) respectively. The energy band gap of untreated sample deposited at -0.7 V and -0.8 V was found to be ~ 1.05 eV and 1.07 eV respectively. However, upon heat treatment at 80°C for an hour the values of bandgap was reduced to 0.9 eV to 1.0 eV due to the enhancement in particle size as well as the formation of polycrystalline CIT material. The estimated

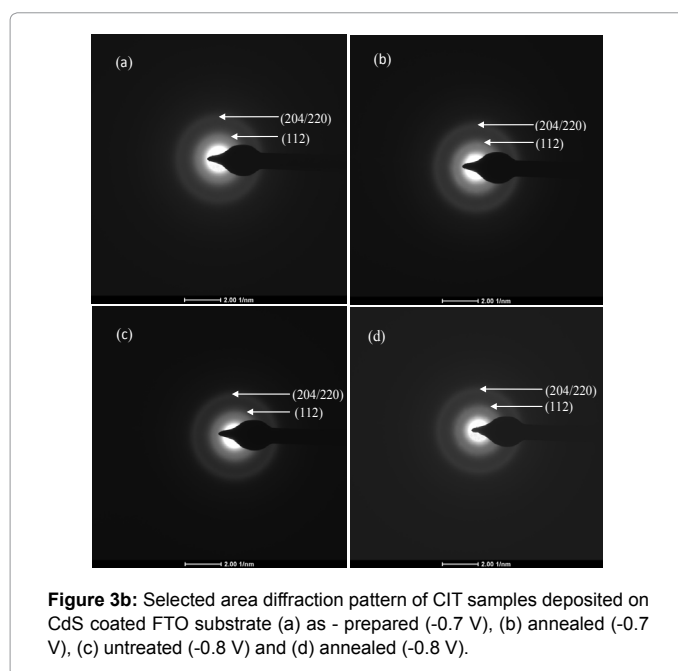
band gap values for untreated and heat treated CIT samples agreed well with the value reported by Mise [14].

### Field Emission Scanning Electron Microscopy (FESEM)

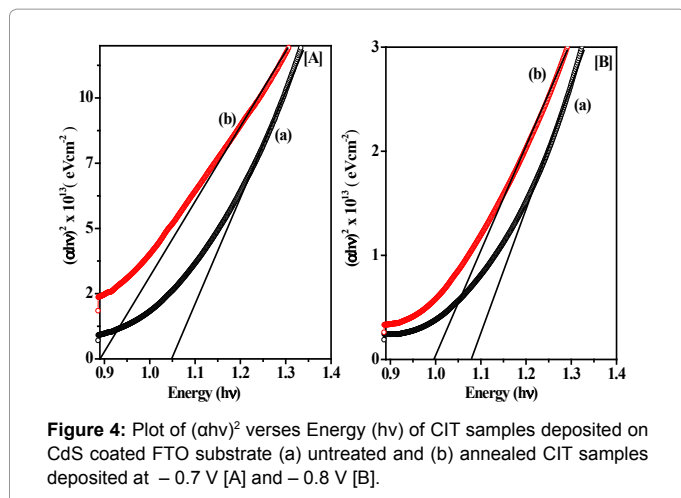
The high magnified surface topographical images recorded for as prepared and heat treated sample is depicted in Figure 5. Compact and adherent thin films were deposited onto CdS substrate with almost uniform grain growth. The morphology of the sample upon thermal heat treatment was observed nearly similar as that of the untreated sample due to the low temperature (80°C) heat treatment. Noticeable change was not expected in the morphology, however the improvement



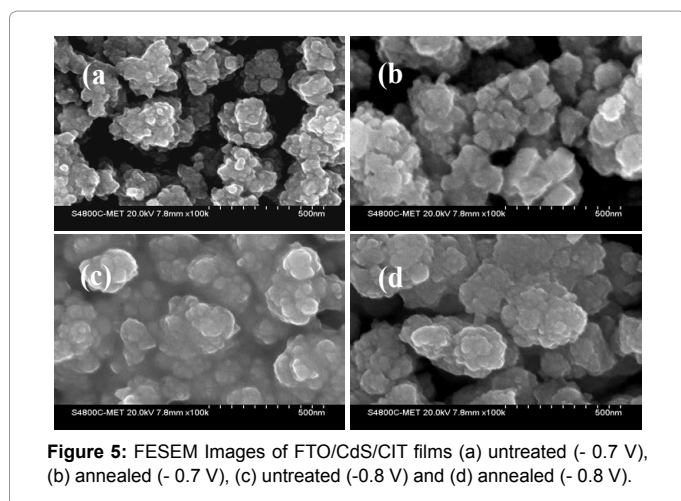
**Figure 3a:** HRTEM micrographs of CIT samples deposited on CdS coated FTO substrate (a) untreated (-0.7 V), (b) annealed (-0.7 V), (c) untreated (-0.8 V) and (d) annealed (-0.8 V). A magnified region of annealed samples (b) and (d) shows fringing pattern.



**Figure 3b:** Selected area diffraction pattern of CIT samples deposited on CdS coated FTO substrate (a) as - prepared (-0.7 V), (b) annealed (-0.7 V), (c) untreated (-0.8 V) and (d) annealed (-0.8 V).



**Figure 4:** Plot of  $(ahv)^2$  versus Energy (hv) of CIT samples deposited on CdS coated FTO substrate (a) untreated and (b) annealed CIT samples deposited at  $-0.7$  V [A] and  $-0.8$  V [B].



**Figure 5:** FESEM Images of FTO/CdS/CIT films (a) untreated ( $-0.7$  V), (b) annealed ( $-0.7$  V), (c) untreated ( $-0.8$  V) and (d) annealed ( $-0.8$  V).

Deposition Potential (Volts)	Sample Condition	Elemental Composition in At. %			Cu/In ratio
		Cu	In	Te	
-0.7	Untreated	19.57	23.05	57.38	0.85
	Heat treated	18.98	25.29	55.73	0.75
-0.8	Untreated	17.81	29.56	52.62	0.60
	Heat treated	22.28	26.28	51.44	0.84

**Table 2:** A summary of elemental composition of CIT samples deposited on CdS substrate.

in the crystallinity noticed upon heat treatment is associated with an enhancement in compactness and reducing the grain boundaries with filling of Te and  $In_xTe_y$  related compound. Figures 5a and 5c are recorded for the untreated samples electrodeposited at  $-0.7$  V and  $-0.8$  V and their corresponding heat treated FESEM images are depicted in Figures 5b and 5d respectively.

A cauliflower like morphology with small visible voids can be clearly seen for the untreated CIT sample deposited at  $-0.7$  V. The voids are due to the large cluster of size  $\sim 400$ - $500$  nm deposited on the surface. Upon heat treatment (Figure 5c) the voids were diminished and a good uniform surface with diffused grains can be seen. On the contrary, well adherent film was deposited at  $-0.8$  V onto CdS substrates. The large clusters of size  $\sim 500$  nm –  $1000$  nm was deposited at higher cathodic potential  $-0.8$  V probably due to higher cathodic current.

## Energy Dispersive Spectroscopy (EDS)

The elemental analysis was determined using the EDS assembly attached with TEM. A summary of the elemental atomic concentration in percentage obtained by EDS analysis is summarized in Table 2. The contents of Cu and Te were found to be more at  $-0.7$  V as a counter part of sample deposited at  $-0.8$  V. As the equilibrium potentials of Cu and Te are more positive than that of In, the deposition rate of Cu and Te was exceeded more at low cathodic potentials. However, nearly stoichiometric CIT thin films were obtained upon the thermal heat treatment on the sample electrodeposited at  $-0.8$  V. Upon heat treatment the Cu/In ratio for the sample deposited at  $-0.7$  V was found to be decreased, whereas, the converse is true for the sample deposited for  $-0.8$  V.

## Hall measurement

Hall measurement set up with constant magnetic field  $0.54$  T and probe current  $10$  mA was employed to determine the electrical parameters hall-coefficient, bulk concentration, mobility, conductivity and resistivity. The average Hall-coefficient for all samples (untreated and heat treated) was positive, confirms the p-type conductivity. The carrier concentration calculated for all samples was of the order of  $10^{19}$   $cm^{-3}$ , which is slightly higher probably due to the higher concentration of Indium (In). As expected the conductivity ' $\sigma$ ' and mobility ' $\mu$ ' were found to be increased upon heat treatment due to recrystallization of the material as well as the reduction of the grain boundaries. However, the mobility ' $\mu$ ' is low as compare to other chalcopyrite materials [30], II-VI semiconductors [31] and silicon [32]. The values of the electrical parameters obtained by Hall measurement are tabulated in Table 3. Mobility values  $\sim 3$   $cm^2/Vs$  and  $1.5$   $cm^2/Vs$  were reported for p-type  $CuIn_3Te_5$  and  $CuIn_2Te_3$  respectively by Wasim [33]. Prabhukanthan [10] has reported  $\sim 63$   $cm^2/Vs$  mobility for as-deposited and  $105$   $cm^2/Vs$  for annealed CIT single crystal grown by chemical vapor transport. In the present data low mobility is due to the higher carrier concentration of the charge carriers. The mobility may be further improved with optimization of heat treatment conditions and surface treatment procedures.

## Current density-Voltage (J-V) characteristics

The highly polycrystalline CIT thin films were obtained upon the heat treatment for the sample grown at  $-0.8$  V versus Ag/AgCl reference, therefore we have prepared the solar cell for  $-0.8$  V. Prior to metal contact the samples were etched in NaCN solution ( $0.2$  wt%) for  $90$  seconds. A typical dark and illuminated photo current density - voltage (J-V) characteristics is shown in Figure 6. Illuminated curve is measured with  $1.5$  AM ( $100$   $mW\ cm^{-2}$ ). The potential barrier,  $\Phi_b$  was calculated under dark condition by the following equation [34];

$$\Phi_b = -\frac{kT}{q} \ln \left( \frac{A^{**} T^2}{J_s} \right) \quad (2)$$

where, ' $\Phi_b$ ' is the barrier height, ' $k$ ' is the Boltzmann's constant, ' $T$ ' is the temperature, ' $q$ ' is the charge on electron, ' $A^{**}$ ' is the effective Richardson's constant for CIT material and ' $J_s$ ' is the reverse saturation current density. The barrier height ' $\Phi_b$ ' was found to  $0.49$  eV. The solar cell exhibits the short circuit current density ( $J_{sc}$ ),  $40.75$   $mA/cm^2$ ; open circuit voltage ( $V_{oc}$ ),  $255$  mV; fill factor (FF),  $41\%$  and power conversion efficiency ( $\eta$ ),  $4.01\%$ . The low measured values of solar cell are due to the high Cu/In ratio,  $0.84$ . The value of Cu/In ratio has to be around  $0.26$ - $0.30$  [6,14] to prepare the high efficiency solar cells. Since the band gap of the CIT is in the IR region, the short circuit current

Deposition Potential (V)	Sample condition	Concentration (cm <sup>-3</sup> )	Mobility (cm <sup>2</sup> /Vs)	Conductivity (1/Ω cm)	Resistivity (Ω cm)	Average Hall Coefficient cm <sup>3</sup> /C
-0.7	Untreated	7.55 × 10 <sup>19</sup>	23.06	2.78 × 10 <sup>2</sup>	3.58 × 10 <sup>-3</sup>	8.27 × 10 <sup>-2</sup>
	Heat treated	6.48 × 10 <sup>19</sup>	29.44	3.05 × 10 <sup>2</sup>	3.27 × 10 <sup>-3</sup>	9.64 × 10 <sup>-2</sup>
-0.8	Untreated	8.36 × 10 <sup>19</sup>	18.40	2.46 × 10 <sup>2</sup>	4.05 × 10 <sup>-3</sup>	7.46 × 10 <sup>-2</sup>
	Heat treated	7.32 × 10 <sup>19</sup>	24.13	2.83 × 10 <sup>2</sup>	3.53 × 10 <sup>-3</sup>	8.52 × 10 <sup>-2</sup>

Table 3: Transport properties of the as-deposited and heat treated CIT samples.

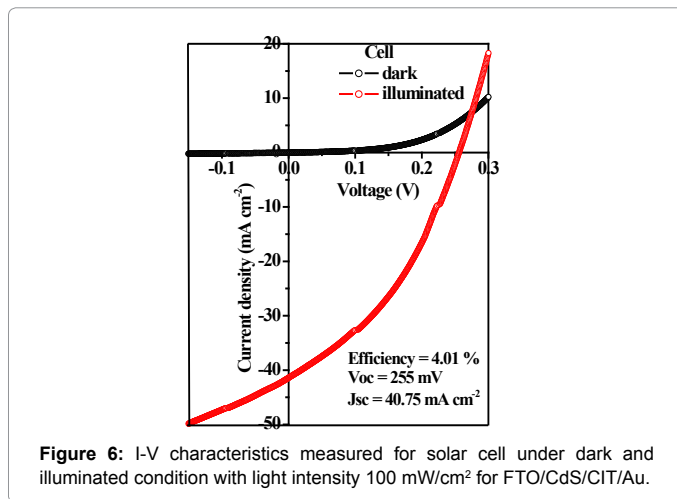


Figure 6: I-V characteristics measured for solar cell under dark and illuminated condition with light intensity 100 mW/cm<sup>2</sup> for FTO/CdS/CIT/Au.

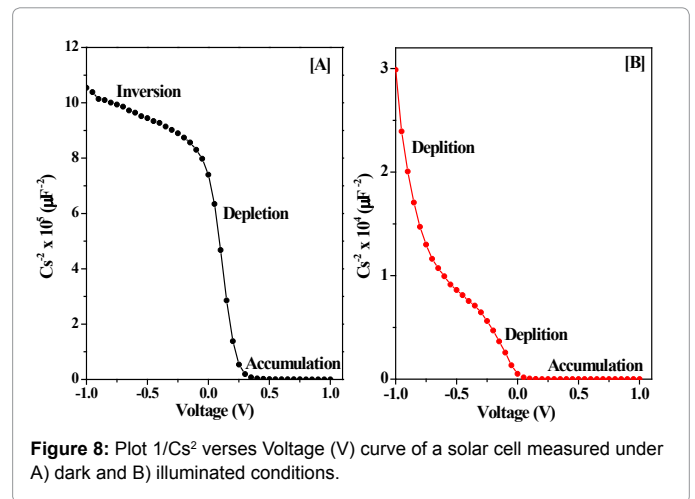


Figure 8: Plot 1/Cs<sup>2</sup> versus Voltage (V) curve of a solar cell measured under A) dark and B) illuminated conditions.

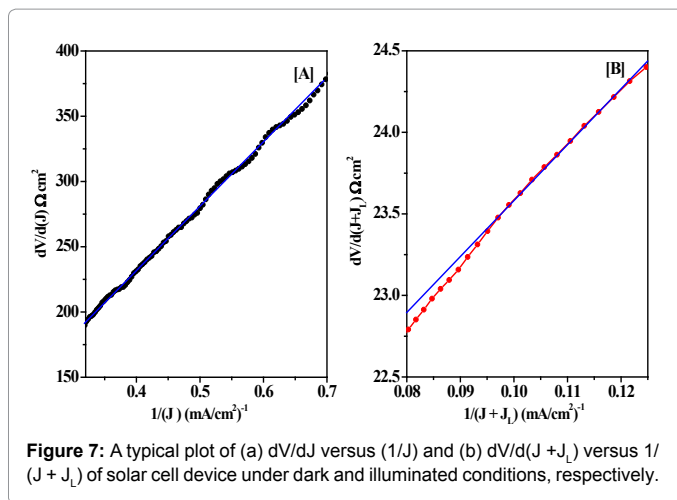


Figure 7: A typical plot of (a) dV/dJ versus 1/J and (b) dV/d(J + J<sub>0</sub>) versus 1/(J + J<sub>0</sub>) of solar cell device under dark and illuminated conditions, respectively.

density has almost close to the theoretical value but V<sub>oc</sub> is found to low could be due to the recombination centers present in the space charge region. The series resistance “R<sub>s</sub>” and ideality factor ‘n’ for cell under dark and illumination condition were deduced from the following diode equation [35];

$$J = J_0 \exp \left[ \frac{q}{nk_B T} (V - JR_s) \right] - J_L \quad (3)$$

The intercept and slope of dV/dJ versus 1/J and dV/d(J + J<sub>0</sub>) versus  $\left( \frac{1}{J + J_0} \right)$  depicted in Figure 7 give the values of series resistance, ‘R<sub>s</sub>’ and ideality factor, ‘n’, respectively. The values of series resistance and ideality factor 193 Ω, 22.9 Ω and 4.84, 3.21 were obtained under dark and illuminated conditions, respectively. Both series resistance (R<sub>s</sub>) and ideality factor (n) can be further reduced with optimization of heat treatment temperature, post –surface treatment and the duration of heating and etching of the sample.

### Capacitance-Voltage measurement

The capacitance – voltage (C<sub>s</sub><sup>-2</sup> versus V) measurement of the above solar cell was studied under dark and illuminated condition for frequency 100 kHz and shown in Figure 8. It is observed that the drift potential/flat band potential under dark is found to ~ 260 mV. Three regions i.e inversion, depletion and accumulation are clearly observed under dark condition. The carrier concentration under dark was calculated by following equation (4) [34] and found to be ~ 1.4 × 10<sup>19</sup> cm<sup>-3</sup> which agrees well with the charge carriers obtained from Hall Effect measurements.

$$N_A = \frac{2}{\epsilon_s q} \left[ \frac{1}{\left( \frac{d(1/C^2)}{dV} \right)} \right] \quad (4)$$

where ‘N<sub>A</sub>’ is the acceptor concentration, ‘q’ is the charge on electron and ‘ε<sub>s</sub>’ is the permittivity of semiconductor. The diffusion of charge carriers begins beyond 0.3 V under dark condition. Under illumination the clear inversion was not observed however, the device has found to be depleted two times showing the dominance of drift mechanism under reverse bias. A diffusion of charge carriers begins around 0.06 V shows the accumulation of charge carriers. It seems that device was found to be accumulated throughout the forward region indicates that forward current is due to the diffusion mechanism [6,36]. The carrier concentration under illumination is ~ 2.9 × 10<sup>21</sup> cm<sup>-3</sup>. The increase in charge carriers could be due to the generation of charge carriers under illumination.

### Conclusion

Cathodic electrodeposition technique was successfully employed to deposit the CuInTe<sub>2</sub> thin films. Highly polycrystalline CIT films were obtained upon heat treatment at 80°C deposited for - 0.8 V. The most

prominent A<sub>1</sub> mode observed from the Raman spectra at 122 cm<sup>-1</sup> in the heat treated samples confirms the formation of chalcopyrite structure. The HRTEM images exhibits clear lattice fringes upon heat treatment. The energy band gap of CIT samples was found to be ~ 1.05 eV and 1.07 eV for the deposited at -0.7 V and -0.8 V, respectively, which was reduced to 0.9 eV to 1.0 eV upon heat treatment. Compact, void free and well adherent layers with cauliflower like morphology of grain size ~ 500-1000 nm were deposited at -0.8 V. Cu/In ratio determined from EDS data was less than unity, which is suitable for high efficiency solar cells. The conductivity 'σ' and mobility 'μ' was found to be increased upon heat treatment due to re-crystallization and reduction in grain boundaries. P-type conductivity of CIT sample was confirmed with hall coefficient measurement. A typical solar cell investigated under illumination (1.5 AM) measured short circuit current density, (J<sub>sc</sub>), 40.75 mA/cm<sup>2</sup>; open circuit voltage (V<sub>oc</sub>), 255 mV; fill factor (FF), 41% and power conversion efficiency (η), 4.01%. This low-temperature heat treatment procedure could be advantageous for the fabrication of CIT solar cells onto flexible substrates. The device performance can be improved by controlling the concentration of depositing species, improving the heterojunction interface and changing the etching parameters.

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